

4-[[*(1E)*-(3,5-Dibromo-2-hydroxyphenyl)methylene]-amino]-1,5-dimethyl-2-phenyl-1,2-dihydro-3*H*-pyrazol-3-one

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The crystal structure of the title compound, $C_{18}H_{15}Br_2N_3O_2$, shows a strong intramolecular O—H···N hydrogen bond [$N\cdots O = 2.609(4)$ Å, $O—H = 0.90$ Å, $H\cdots N = 1.80$ Å and $O—H\cdots N = 148^\circ$], which leads to the existence of a phenol-imine tautomer.

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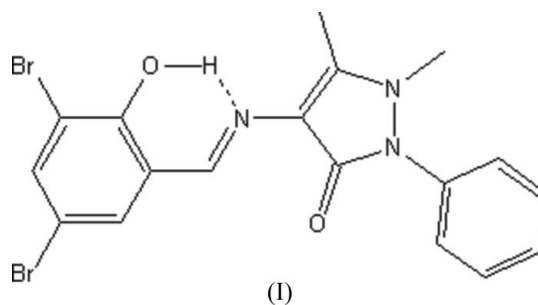
Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(C—C) = 0.004$ Å
 R factor = 0.032
 wR factor = 0.078
Data-to-parameter ratio = 13.4

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Comment

The Schiff bases derived from salicylaldehyde have been much studied because of their properties, such as tautomeric (Salman *et al.*, 1991), fluorescent (Morishige *et al.*, 1980), and thermo- and photochromic properties (Barbara *et al.*, 1980; Cohen *et al.*, 1964). In a search for new analytical reagents, we have synthesized some compounds of substituted salicylaldehyde with 4-aminoantipyrine. We report here the synthesis and crystal structure of the title compound, (I).



All bond distances and angles are normal and agree with the corresponding values found in a similar compound 4-[(2-hydroxy-3-methoxybenzylidene)amino]-1,5-dimethyl-2-phenyl-1,2-dihydro-3*H*-pyrazol-3-one (Diao *et al.*, 2005). There is an intramolecular O—H···N hydrogen bond (Table 2); the compound is, therefore, in the phenol-imine form, as in 4-[[*(1Z)*-2-hydroxyphenyl)methylene]amino]-1,5-dimethyl-2-phenyl-1,2-dihydro-3*H*-pyrazol-3-one [$N1\cdots O1 = 2.607(3)$ Å, $O1—H1 = 0.97(3)$ Å, $H1\cdots N1 = 1.71(3)$ Å and $O1—H1\cdots N1 = 153(2)^\circ$; Hökelek *et al.*, 2001].

Experimental

3,5-Dibromosalicylaldehyde was synthesized according to the published method (Brewster, 1924). Ethanol solutions of 3,5-dibromosalicylaldehyde (10 mmol, 2.80 g) and 4-aminoantipyrine (10 mmol, 2.03 g) were mixed and refluxed on a water bath for 5 h. The precipitate was filtered off and recrystallized from methanol (yield 81%, m.p. 501–502 K). IR (KBr, cm^{-1}): ν_{max} 3410.5, 1659.6, 1591.2, 1446.5, 1363.6, 1290.3, 1135.0, 765.7. 1H NMR (200 MHz, $CDCl_3$): δ 14.42 (1H), 9.71 (1H), 7.22–7.72 (7H), 3.21 (3H), 2.42 (3H).

Crystal data

$C_{18}H_{15}Br_2N_3O_2$
 $M_r = 465.15$
 Monoclinic, $P2_1/n$
 $a = 7.092$ (5) Å
 $b = 8.231$ (5) Å
 $c = 30.487$ (2) Å
 $\beta = 91.725$ (12)°
 $V = 1779$ (2) Å³
 $Z = 4$

$D_x = 1.737$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 3452 reflections
 $\theta = 2.6$ – 25.3 °
 $\mu = 4.57$ mm⁻¹
 $T = 293$ (2) K
 Block, orange
 $0.24 \times 0.19 \times 0.12$ mm

Data collection

Siemens SMART CCD area detector diffractometer
 ω and φ scans
 Absorption correction: multi-scan (SADABS; Bruker 2002)
 $T_{min} = 0.367$, $T_{max} = 0.578$
 9143 measured reflections

3224 independent reflections
 2631 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.034$
 $\theta_{max} = 25.3$ °
 $h = -8 \rightarrow 8$
 $k = -9 \rightarrow 9$
 $l = -33 \rightarrow 36$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.032$
 $wR(F^2) = 0.078$
 $S = 1.03$
 3224 reflections
 239 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.035P)^2 + 0.3014P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} = 0.001$
 $\Delta\rho_{max} = 0.41$ e Å⁻³
 $\Delta\rho_{min} = -0.34$ e Å⁻³
 Extinction correction: SHELXL97
 Extinction coefficient: 0.0018 (4)

Table 1

Selected geometric parameters (Å, °).

Br1—C3	1.900 (3)	N1—C7	1.292 (4)
Br2—C5	1.889 (3)	N1—C8	1.393 (4)
O1—C6	1.344 (3)	N2—N3	1.414 (3)
O2—C10	1.228 (3)	N3—C10	1.409 (3)
C7—N1—C8	120.4 (3)	O1—C6—C5	120.3 (3)
C2—C1—C7	118.9 (3)	O1—C6—C1	121.9 (2)

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1—H1A \cdots N1	0.90	1.80	2.609 (4)	148
C7—H7A \cdots O2	0.93	2.26	2.958 (4)	131
C18—H18A \cdots O2	0.93	2.46	2.882 (4)	108

The hydroxy H atom (H1A) was positioned from a difference map, refined for several cycles and then fixed at a distance of 0.90 Å; the methyl H atoms on C11 were located in a Fourier synthesis and refined freely. The remaining H atoms were positioned geometrically and treated as riding, at distances of 0.93 (CH) and 0.96 Å (CH₃) [$U_{iso}(H) = 1.2U_{eq}(C)$ or $1.5U_{eq}(C_{methyl},O)$].

Data collection: SMART (Bruker 2002); cell refinement: SAINT (Bruker 2002); data reduction: SAINT and SHELXTL (Bruker

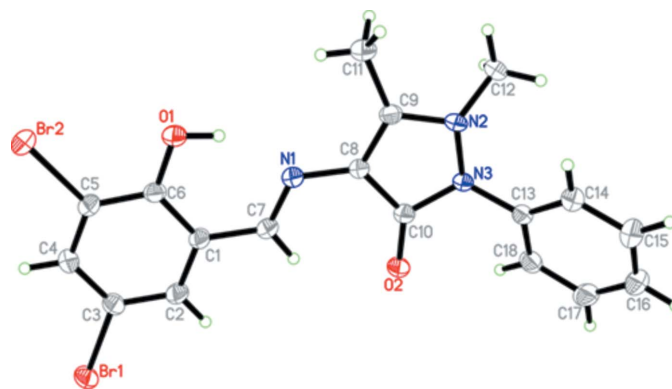


Figure 1

The structure of (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme.

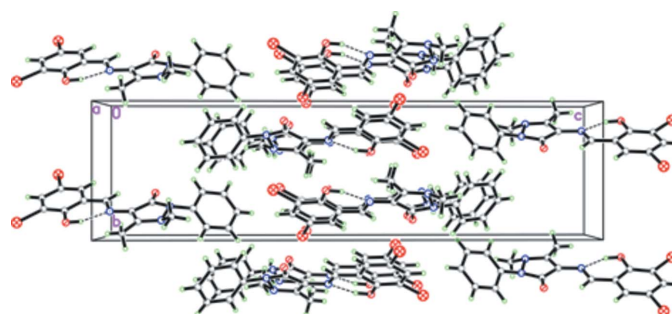


Figure 2

The packing of (I), viewed down the a axis. Dashed lines indicate hydrogen bonds.

2002); program(s) used to solve structure: SHELXTL; program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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